SPE RESPONSE FOR CERTIFICATE OF CORRECTION

		Paper No.:1230	05
DATE	: December 30, 2005		
TO SPE OF	: ART UNIT 1725	- .	
SUBJECT	: Request for Certificate of Correct	ction on Patent No.: 6,803,337	
A response is	requested with respect to the accomp	panying request for a certificate of correction.	
Certificates	lete this form and return with file, of Correction Branch - PK 3-910 n 7590 - Tel. No. 305-8201		
With respect to the change(s) requested, correcting Office and/or Applicant's errors, should the patent read as shown in the certificate of correction? No new matter should be introduced, nor should the scope or meaning of the claims be changed.			
Thank You F	For Your Assistance	Certificates of Correction Branch	
The request for issuing the above-identified correction(s) is hereby: Note your decision on the appropriate box.			
⊠ Ар	proved	All changes apply.	
□ Ар	proved in Part	Specify below which changes do not apply.	
☐ De	nied	State the reasons for denial below.	
Comments:			
		SPE: Patrick Ryan Art Unit 172	5

CERTIFICATE OF CORRECTION

PATENT NO.:

6,803,337

DATED:

October 12, 2004

INVENTOR(S):

Laura ZANIBELLI, et al.

It is certified that an error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

Column 9, Claim 1 should read:

A catalytic composition having a surface area $\geq 380 \text{ m}^2/\text{g}$ and a pore volume $\geq 1.10 \text{ cm}^3/\text{g}$ consisting of:

one or more oxides as a carrier and

a catalyst consisting of a beta zeolite, cobalt, a metal of Group VI B wherein said zeolite is present in a quantity ranging from 70 to 90% by weight, the weight percentage of cobalt varies from 1 to 10% with respect to the total weight of the catalyst, the weight percentage of the metal of Group VI B varies from 4 to 20% with respect to the total weight of the catalytic composition, and wherein cobalt and the metal of Group VI B are present in a molar ratio not greater than 2.

Column 11, Claim 22 should read:

The process of Claim 21, wherein the catalyst has a surface area of $\geq 380 \text{ m}^2/\text{g}$ and a pore volume $\geq 1.10 \text{ cm}^3/\text{g}$.

Column 13, Claim 38 should read:

The process of claim 37, wherein the catalyst has a surface area $\geq 380 \text{ m}^2/\text{g}$ and a pore volume $\geq 1.10 \text{ cm}^3/\text{g}$.

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6,803,337

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October 12, 2004

INVENTOR(S):

Laura ZANIBELLI, et al.

It is certified that an error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

Column 13, Claim 39 should read:

A catalytic composition having a surface area $\geq 380 \text{ m}^2/\text{g}$ and a pore volume \geq

1.10 cm³/g consisting of

one or more oxides as a carrier and

a catalyst consisting of a beta zeolite, cobalt, a metal of Group VI B wherein said zeolite is present in a quantity ranging from 70 to 90% by weight, the weight percentage of cobalt varies from 1 to 10% with respect to the total weight of the catalyst, the weight percentage of the metal of Group VI B varies from 4 to 20% with respect to the total weight of the catalytic composition, and wherein cobalt and the metal of Group VI B are present in a molar ratio not greater than 2, prepared with a process consisting of the following steps:

- a) an alcoholic dispersion is prepared, containing a soluble salt of cobalt, beta zeolite and one or more organic compounds capable of generating the supporting oxide or oxides;
- b) an aqueous solution is prepared containing a soluble salt of the metal of group VI B and, optionally, tetrealkylammonium hydroxide having the formula R4NOH;
 - c) the alcoholic dispersion and the aqueous dispersion are mixed and a gel is obtained;
 - d) ageing the gel at a temperature ranging from 10 to 40°C;
 - e) drying the gel; and
 - f) calcinating the gel.

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INVENTOR(S):

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It is certified that an error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

Column 9, Claim 1 should read:

A catalytic composition having a surface area ≥ 380 m²/g and a pore volume ≥ 1.10 cm³/g consisting of:

one or more oxides as a carrier and

a catalyst consisting of a beta zeolite, cobalt, a metal of Group VI B wherein said zeolite is present in a quantity ranging from 70 to 90% by weight, the weight percentage of cobalt varies from 1 to 10% with respect to the total weight of the catalyst, the weight percentage of the metal of Group VI B varies from 4 to 20% with respect to the total weight of the catalytic composition, and wherein cobalt and the metal of Group VI B are present in a molar ratio not greater than 2.

Column 11, Claim 22 should read:

The process of Claim 21, wherein the catalyst has a surface area of $\geq 380 \text{ m}^2/\text{g}$ and a pore volume $\geq 1.10 \text{ cm}^3/\text{g}$.

Column 13, Claim 38 should read:

The process of claim 37, wherein the catalyst has a surface area $\geq 380 \text{ m}^2/\text{g}$ and a pore volume $\geq 1.10 \text{ cm}^3/\text{g}$.

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INVENTOR(S):

Laura ZANIBELLI, et al.

It is certified that an error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

Column 13, Claim 39 should read:

A catalytic composition having a surface area ≥ 380 m²/g and a pore volume ≥

1.10 cm³/g consisting of

one or more oxides as a carrier and

a catalyst consisting of a beta zeolite, cobalt, a metal of Group VI B wherein said zeolite is present in a quantity ranging from 70 to 90% by weight, the weight percentage of cobalt varies from 1 to 10% with respect to the total weight of the catalyst, the weight percentage of the metal of Group VI B varies from 4 to 20% with respect to the total weight of the catalytic composition, and wherein cobalt and the metal of Group VI B are present in a molar ratio not greater than 2, prepared with a process consisting of the following steps:

- a) an alcoholic dispersion is prepared, containing a soluble salt of cobalt, beta zeolite and one or more organic compounds capable of generating the supporting oxide or oxides;
- b) an aqueous solution is prepared containing a soluble salt of the metal of group VI B and, optionally, tetrealkylammonium hydroxide having the formula R4NOH;
 - c) the alcoholic dispersion and the aqueous dispersion are mixed and a gel is obtained;
 - d) ageing the gel at a temperature ranging from 10 to 40°C;
 - e) drying the gel; and

f) calcinating the gel.

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